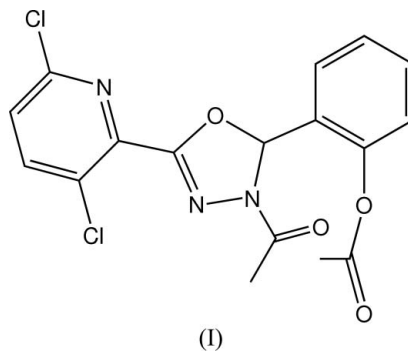


Qing-Bao Song,^{a*} Jie Zhang^a and
Edward R. T. Tiekink^{b*}^aThe State Key Laboratory Breeding Base of
Green Chemistry-Synthesis Technology, College
of Chemical Engineering and Materials Science,
Zhejiang University of Technology, Hangzhou
310014, People's Republic of China, and^bDepartment of Chemistry, The University of
Texas at San Antonio, One UTSA Circle, San
Antonio, Texas 78249-0698, USACorrespondence e-mail: qbsong6@163.com,
edward.tiekink@utsa.edu

Key indicators

Single-crystal X-ray study
 $T = 173\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.040
 wR factor = 0.116
Data-to-parameter ratio = 16.7For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2-[3-Acetyl-5-(3,6-dichloropyridin-2-yl)-
2,3-dihydro-1,3,4-oxadiazol-2-yl]phenyl
acetateThe title compound, $\text{C}_{17}\text{H}_{13}\text{Cl}_2\text{N}_3\text{O}_4$, shows the substituted
pyridine ring to be coplanar with the oxadiazoline ring but the
third ring to be normal to this. Chains mediated by $\text{C}-\text{H}\cdots\text{O}$
interactions are linked to two other chains *via* $\text{C}-\text{H}\cdots\text{Cl}$
interactions and to another two chains *via* $\text{C}-\text{H}\cdots\pi$
interactions.

Comment

Oxadiazolines represent a class of heterocycle that, although
known for over a century, is rather limited in number. These
are partially reduced forms of well known oxadiazoles.
Recently, oxadiazolines have received increased attention
owing to their usefulness as synthetic intermediates (Charmier
et al., 2004; Moustafa, 2003) and promising biological activity.
For example, oxadiazolines are reported to possess anti-
tumour (Chimirri *et al.*, 1996), anti-HIV (Chimirri *et al.*, 1994),
antifungal (Singh & Hasan, 2002), anti-inflammatory
(Tinperciuc *et al.*, 1999) and anticonvulsant (Dogan *et al.*,
1998) properties. 3-Acetyl-2,5-disubstituted-1,3,4-oxadiazolines
are also known to exhibit antimicrobial activity (Hassan
et al., 1983; Khalil *et al.*, 1993). It was in this context that the
title compound, (I), was prepared.The molecule of (I) (Fig. 1) is twisted about the $\text{C}2-\text{C}10$
bond, as seen in the $\text{O}1/\text{C}2/\text{C}10/\text{C}11$ torsion angle of
 $58.02(19)^\circ$. This twist most likely arises to relieve the putative
strain between the $\text{N}3$ -acyl and $\text{C}11$ -acetyl groups. The
consequence of this twist is that the $\text{C}10-\text{C}15$ ring is almost
normal to the oxadiazoline ring, as seen in the dihedral angle
between their least-squares planes of $89.11(8)^\circ$. By contrast,
the oxadiazoline is effectively coplanar with the $\text{N}1/\text{C}3-\text{C}7$
ring [dihedral angle = $4.58(8)^\circ$]. Selected interatomic param-
eters are collected in Table 1 and these show that the $\text{N}2-\text{C}1$
bond distance corresponds to a double bond, indicating
limited delocalization of π -electron density over the
oxadiazoline ring despite its planarity; maximum deviation ofReceived 9 August 2006
Accepted 21 August 2006

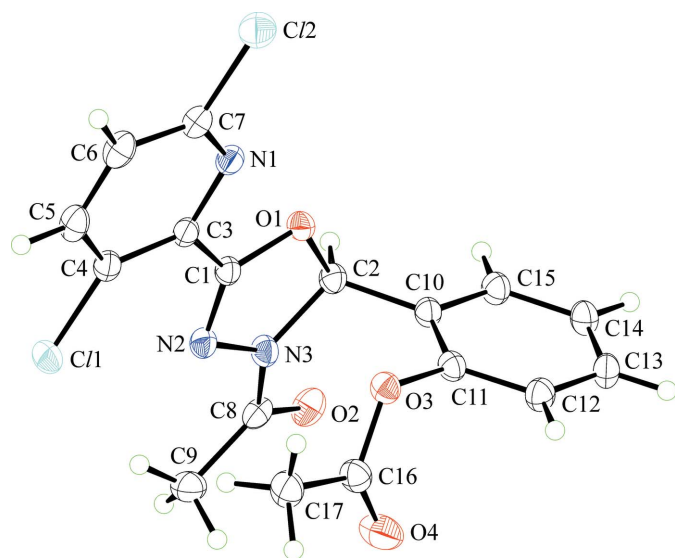


Figure 1
The atom-labelling scheme for (I), showing 50% probability displacement ellipsoids.

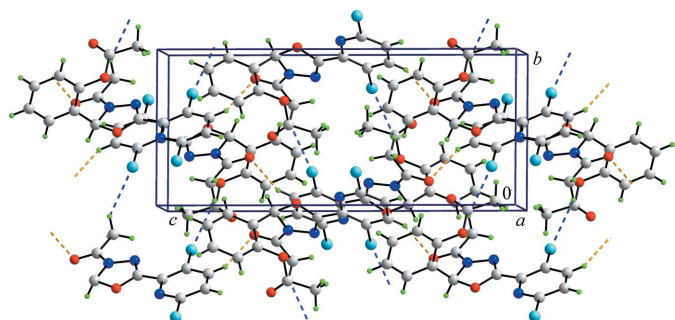


Figure 2
Chain formation in (I), viewed approximately down the *a* axis (Crystal Impact, 2006). Colour code: Cl (cyan), O (red), N (blue), C (grey) & H (green). Gold and blue dashed lines indicate C—H...O and C—H...Cl hydrogen-bonding interactions, respectively.

0.0185 (16) Å for atom C2. The acyl group is coplanar with the oxadiazoline ring, as seen in the N2—N3—C8—O2 torsion angle of 171.00 (14)°, but the acetyl group is normal to the C10—C15 ring, as evidenced by the C10—C11—O3—C16 torsion angle of −109.07 (17)°.

The crystal structure is stabilized by C—H...O, C—H...Cl and C—H... π interactions. A chain along the *c*-axis direction is mediated by C—H...O interactions, highlighted as orange dashed lines in Fig. 2, so that C5—H5...O2ⁱ is 2.29 Å, C5...O2ⁱ is 3.182 (2) Å and the angle at H is 155° [symmetry code: (i) $x, \frac{1}{2} - y, \frac{1}{2} + z$]. These are linked to centrosymmetrically related chains *via* C—H...Cl interactions aligned along the *b*-axis direction and highlighted as blue dashed lines in Fig. 2, with C9—H9^c...Cl1ⁱⁱ of 2.79 Å, C9...Cl1ⁱⁱ of 3.620 (2) Å and angle at H of 143° [symmetry code: (ii) $1 - x, 1 - y, 1 - z$]. Atom N2 does not form an intermolecular interaction but forms an intramolecular C9—H9^b...N2 contact of 2.41 Å, so that C9...N2 is 2.836 (3) Å and the angle at H is 106°. Finally, C—H... π interactions are evident, indi-

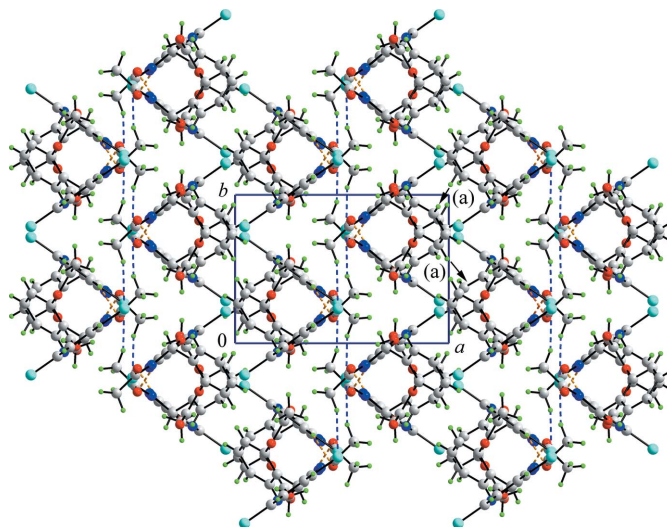


Figure 3
The crystal packing in (I), viewed down the *c* axis. Colour code and hydrogen bonds as for Fig. 2.

cated by '*a*' in Fig. 3, that link a further two chains to the original so that each chain is surrounded by four chains in total. Here, the C12—H...Cgⁱⁱⁱ distance (Cg is the centroid of ring C10—C15) is 2.79 Å, with an angle of 172° at H [symmetry code: (iii) $-x, \frac{1}{2} + y, \frac{1}{2} - z$].

Experimental

Compound (I) was prepared in accord with the literature procedure (Yale *et al.*, 1953). Colourless crystals were obtained by slow evaporation of an ethyl acetate solution of (I) after 4 d at room temperature (m.p. 446 K).

Crystal data

C₁₇H₁₃Cl₂N₃O₄
M_r = 394.20
 Monoclinic, *P*2₁/*c*
a = 11.839 (4) Å
b = 7.917 (3) Å
c = 19.255 (7) Å
 β = 105.658 (6)°
V = 1737.6 (10) Å³

Z = 4
D_x = 1.507 Mg m^{−3}
 Mo *K* α radiation
 μ = 0.40 mm^{−1}
T = 173 (2) K
 Plate, colourless
 0.35 × 0.35 × 0.10 mm

Data collection

Rigaku AFC12K/SATURN724
 diffractometer
 ω scans
 Absorption correction: multi-scan
 (ABSCOR; Higashi, 1995)
T_{min} = 0.851, *T_{max}* = 1

45244 measured reflections
 3956 independent reflections
 3911 reflections with *I* > 2 σ (*I*)
R_{int} = 0.030
 θ_{\max} = 27.5°

Refinement

Refinement on *F*²
R[*F*² > 2 σ (*F*²)] = 0.040
wR(*F*²) = 0.116
S = 1.16
 3956 reflections
 237 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0598P)^2 + 0.6091P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.28 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.29 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

C1—C4	1.7245 (17)	O4—C16	1.186 (2)
C12—C7	1.7423 (18)	N1—C7	1.311 (2)
O1—C1	1.3675 (17)	N1—C3	1.348 (2)
O1—C2	1.4414 (17)	N2—C1	1.282 (2)
O2—C8	1.218 (2)	N2—N3	1.3871 (17)
O3—C16	1.373 (2)	N3—C8	1.363 (2)
O3—C11	1.4037 (18)	N3—C2	1.4727 (19)
C1—O1—C2	106.63 (11)	N2—N3—C8	124.06 (13)
C1—N2—N3	104.40 (12)	C2—N3—C8	124.02 (13)
N2—N3—C2	111.64 (12)		

The C-bound H atoms were included in the riding-model approximation with C—H = 0.95–0.98 Å, and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{methyl C})$ or $1.2U_{\text{eq}}(\text{C})$.

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *DIAMOND* (Crystal Impact, 2006); software used to prepare material for publication: *SHELXL97*.

This project was sponsored by the State Innovation Foundation (People's Republic of China).

References

- Altomare, A., Casciarano, M., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Charmier, M. A. J., Haukka, M. & Pombeiro, A. J. L. (2004). *Dalton Trans.* pp. 2741–2745.
- Chimirri, A., Grasso, S., Monforte, A. M., Monforte, P., Zappala, M. & Carotti, A. (1994). *Il Farmaco*, **49**, 509–511.
- Chimirri, A., Grasso, S., Monforte, A. M., Rao, A. & Zappala, M. (1996). *Il Farmaco*, **51**, 125–129.
- Crystal Impact (2006). *DIAMOND*. Version 3.1. Crystal Impact GbR, Bonn, Germany.
- Dogan, H. N., Duran, A., Rollas, S., Sener, G., Armutak, Y. & Keyer-Uysal, M. (1998). *Med. Sci. Res.* **26**, 755–758.
- Hassan, E., Al-Ashmawi, M. I. & Abd El-Fattah, B. (1983). *Pharmazie*, **38**, 833–835.
- Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
- Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Khalil, M. A., El-Sayed, O. A. & El-Shamy, H. A. (1993). *Arch. Pharm. (Weinheim)*, **326**, 489–492.
- Moustafa, A. H. (2003). *Synthesis*, pp. 837–840.
- Rigaku (2005). *CrystalClear* User Manual, Rigaku/MSI Inc., The Woodlands, Texas, USA.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Singh, C. P. & Hasan, H. (2002). *J. Ind. Counc. Chem.* **19**, 46–49.
- Tinperciuc, B., Parvu, A., Palage, M., Oniga, O. & Ghiran, D. (1999). *Farmacia (Bucharest)*, **47**, 77–84.
- Yale, H. L., Losee, K., Martins, J., Holsing, M., Perry, F. M. & Bernstein, J. (1953). *J. Am. Chem. Soc.* **75**, 1933–1942.